

PS#5 PHYS 310 Due Friday May 3, in class

#1. Dinner Estimate: In 2011 DH organized the second of three Physics of Sustainable Energy conferences at Berkeley – I helped a little. The banquet speech not surprisingly was [Art Rosenfeld](#), who gave half of his presentation time to [Richard Muller](#). Funded by big oil interests such as the [Koch Brothers](#), Richard promised to *correct* the “biased” global warming data with his [BEST](#) model results. Richard had managed to offend many climate scientists with his arrogant dismissal of their work, so it was a little tense as we waited. The guy sitting next to me said, “so would someone figure out if we get more heat from burning something than from the associated GHG emissions.” I responded that the heating from the GHG emissions would continue forever, so the real question is “how long will it take for the GHG-radiative forcing heat to be equal to the combustion heat?” and added that we should find a physicist to do that calculation (this was an American Physical Society Conference). I asked myself, “what would DH do?” I started with the CO₂ emissions of burning coal to release a MJ of thermal energy. Then I calculated the associated % increase in total anthropogenic GHG emissions and corresponding increase in the radiative forcing.

- a) Estimate the total anthropogenic GHG emissions from all time, using Pete’s “1930 triangle method” (Carbon emissions began in 1930 and grew roughly linearly to today... what are present global emissions?. By what portion did your little combustion experiment increase our total emissions?
- b) What would be the increase in radiative forcing due to this increase in GHG?
- c) Can you calculate the rate of GHG thermal heating due to burning this coal, and thereby find the time required to provide the heat equal to that directly given off in the burning?
- d) Can you make an argument that this increase may not be linear and so my estimate is (a little) too big?

#2 You drive to San Francisco and back in a car of your choice – name the car.

- a) How much gas did you use?
- b) How many kg of CO₂ did your trip contribute to the environment?
- c) What is the rate of heat dissipation into the environment if your trip was just an immediate “there and back”. Please put your answer in Watts.
- d) What portion of the earth’s remaining accessible petroleum did your trip use?
- e) By what portion did your trip increase the total anthropogenic GHG?

#3 Calculating your carbon footprint: Please estimate your carbon footprint in Tons of CO₂ per year. Do this by doing rough calculations on your use of:

- a) Car transport – calculate gallons used by you. So if you and your room mate need 200 gallons for the year to commute to Cal Poly from Los Osos, you are responsible for 100 Gallons.
- b) Air transport. What is the mileage of the planes you travel in? How far do you fly? How many people share this fuel? You can say that the planes use kerosene... but it is like gasoline.
- c) Use of Natural Gas and Electricity
- d) Food see below for a paper that will help you.
- e) Clothing and other goods

For the last two categories: The embedded carbon in the things you buy, there is a nice publication with some good graphs for you to look at (in particular Fig #2): <http://escholarship.org/uc/item/55b3r1qj>

#4 Climate Change and You: Chris Jones, at Berkeley has created a spreadsheet-based tool for individuals to estimate all direct and indirect emissions of GHGs in CO₂ equivalent units resulting from their primary energy related choices: transportation, food, housing (including energy use), goods and services, and waste. Go to Chris' website: <http://consumerfootprint.org/> and spend some time exploring to understand the basic logic of what the site. Go to the calculators and fill out the widget that pertains to you, most likely the household widget at the bottom of the calculator page. Fill out the carbon calculator with your individual information for each sector. Please be as honest as possible (I won't publish your names). Go to the Summary tab and save your answers.

- a) What is the ratio of the highest category (e.g. housing, food, etc.) of your emissions to the lowest? What is the ratio of your total emissions to the national average of 20 tons CO₂ per person per year? What is the ratio of your total emissions to the global average of 4.5 tons CO₂ per person per year?
- b) What do you find most surprising about your results? Please explain in a short paragraph.
- c) What lifestyle changes would you have to make in order to emit no more than the global per capita average of 4.5 tons of CO₂? To do this, change the values you entered in the spreadsheet until your total emissions are below 4.5 tons CO₂. Try to make realistic choices. What do the results say about how your lifestyle compares to the lifestyles of the majority of people on the planet? Could you live at or below the global average? Please explain in one short paragraph.
- d) (10 points) What if, starting June 1, 2012, everyone suddenly started living like the average American (American per capita CO₂ emissions: http://en.wikipedia.org/wiki/List_of_countries_by_carbon_dioxide_emissions_per_capita)? How quickly would we reach what many climate scientists consider to be the dangerous level of 500 ppm (parts per million) of carbon dioxide in the atmosphere? State any assumptions you make.
- e) (5 pts) How did your estimate from #3 differ from the output of the calculator above? Can you comment on reasons for the difference?

#5 (less likely to be directly pertinent toward an assessment) Dan Kammen at Berkeley says, "we're not running out of petroleum, we're running on of atmosphere. Please investigate this statement.

- a) In the global stocks and flow energy diagram (this is provided for April 3 class on the class website), please estimate how many years our oil will last if we continue using it at the present rate. Assume that we can use every last drop (impossible – most of it is presently considered not accessible). But our rate of use is increasing.
- b) Estimate the mass of CO₂ emitted into the atmosphere for every gallon of gas you burn, then
- c) estimate the mass of CO₂ in the atmosphere if (when) we did use every last drop of petroleum.
- d) Calculate the total mass of the atmosphere knowing that atmospheric pressure times the surface area of the earth is the force of gravity acting on the entire atmosphere (right?)
- e) Knowing that the molecular mass of the atmosphere is about 29 g/mole, estimate the PPM increase in CO₂ in the atmosphere from burning all the petroleum.
- f) Estimate the increase in the Radiative Forcing from this increase in CO₂.
- g) Estimate the increase in global temperature from this increase in CO₂. You may find it helpful to consider the layer model that DH and I published a few years back: <http://www.aps.org/units/fps/newsletters/200807/hafemeister.cfm> ... But you may not find it helpful too.

#6 (**VERY** pertinent toward and Assessment). From the class slide:

Climate Change. I can describe

- The causes of climate change
- Distinguish CC from Ozone Hole
- Define Radiative Forcing
 - What are units of Radiative Forcing?
 - Relative increase in incoming heat?
 - Total Thermal Power return to Earth?
- Positive and Negative Feedback?
- Tipping point?
- Physical Effects?
- Biological Effects?